



- Combustion is an exothermic reaction that releases energy in the forms of heat and light.
- When a fuel undergoes complete combustion, it releases the maximum amount of energy from the fuel being reacted.
- Complete combustion is usually characterized by a blue flame

![](_page_1_Picture_4.jpeg)

![](_page_2_Picture_1.jpeg)

![](_page_2_Figure_2.jpeg)

- Incomplete combustion is often undesirable because it releases less energy than complete combustion and produces carbon monoxide which is a poisonous gas.
- Incomplete combustion can also produce pure carbon (soot) which is messy and can build up in equipment. (ie: chimneys)
- Incomplete combustion is characterized by an orange coloured flame.

![](_page_3_Picture_4.jpeg)

## Uses of Incomplete Combustion

- For centuries, humans have used incomplete combustion of wood to smoke and preserve meat.
- When wood is burned with limited oxygen, a large amount of smoke is produced which dehydrates the meat and forms a protective layer (pellicle) that resists bacteria growth.

- Gasification is the process of incompletely combusting wood or coal to produce flammable gasses (CO and H<sub>2</sub>) that can be used as a fuel.
- Until the mid 1900's many cities produced and distributed "town gas" which consisted of CO, H<sub>2</sub> and CH<sub>4</sub> from the incomplete combustion of coal. This town gas was used for heating, cooking and lighting until natural gas became commonly available in the 1930's and 1940's.

- During World War II, a shortage gasoline and diesel fuel led many people, and even militaries, to convert vehicles to run on wood gas.
- By 1945, over one million vehicles worldwide had been converted to run on wood gas, including over 500 000 in Germany.

![](_page_5_Picture_3.jpeg)

![](_page_6_Picture_1.jpeg)

![](_page_6_Picture_2.jpeg)